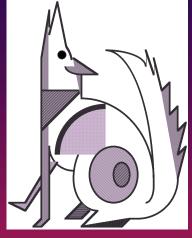
A Career in Thermodynamics, Geochemistry, and Materials Science





Alexandra Navrotsky UC Davis

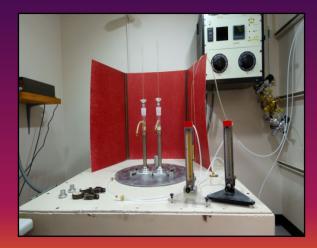
Why I Count Calories for a Living

- They are fascinating
 - Energetics whisper secrets of the strength of chemical bonds
 - Entropies sing of vibrating atoms, moving electrons, and structural disorder
 - Systematics have predictive power
- They pay
 - thermodynamic data are essential to good materials processing
 - Environmental science needs thermodynamics, both for issues of stability and as a starting point for kinetics
 - Mineralogy, petrology, and deep Earth geophysics need thermodynamic data.

Calorimetry Measures

- Heat capacities
- Heats of phase transitions'
- Heats of formation





Commercial Setaram AlexSYS Calorimeter



From these data one calculates

Enthalpies, entropies, free energies

- Solubililities
- Phase diagrams
- Petrologic and geochemical processes
- Materials synthesis and compatibility

Home Runs

- Spinel cation distributions and thermodynamics:
 1967-present
- Lower mantle mineralogy, negative P-T slopes for perovskite forming reactions: 1985-95
- Energy landscapes in zeolites, MOFs, and other porous materials: 1992-present
- Huge shifts in thermodynamics and phase equilibria at the nanoscale: 2008-present
- Other events and ambience in these time periods

Events and Ambience 1965-80

The World

- Emphasis on science and the space race
- Assassinations and riots of 1968
- Oil crisis of 1973
- Major universities simply not hiring women or minorities
- Research funding relatively good

Alex

- Ph.D. U. Chicago 1967
- Postdoc in Germany 1967-68
- Postdoc Penn State 1968-69
- Faculty position ASU 1969-85 a splendid start
- Established research program
 - Spinels
 - Ceramics
 - Minerals
- NSF rotator 1976-77

SPINELS 1967

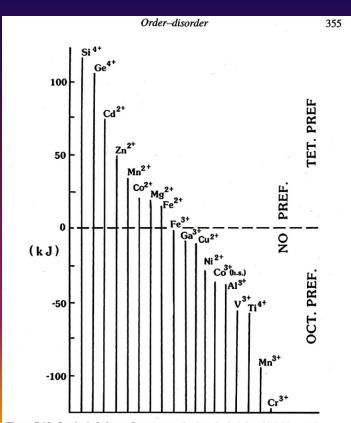
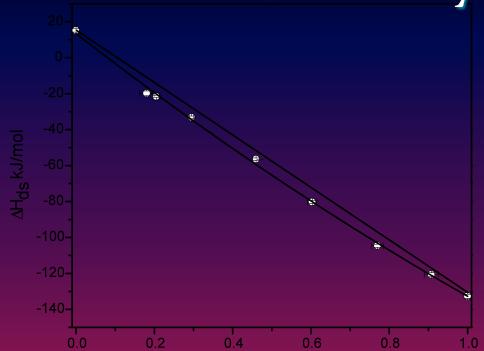


Figure 7.12. Octahedral site preference energies in spinels (after O'Neill and Navrotsky 1984).

Experimental thermodynamic basis of site preference energies and enthalpies of formation of spinels. Crystal field effects are but a small contribution

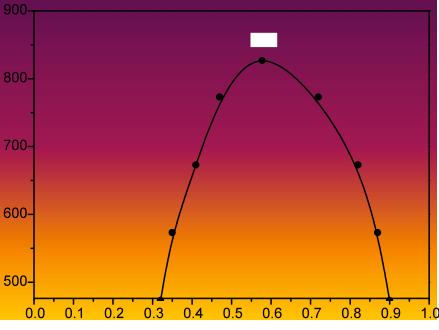
A dream- high pressure silicate spinels

Spinels now: Magnetite-Ulvöspinel Thermodynamics



Calculated solvus in the Fe_3O_4 – Fe_2TiO_4 system. The critical temperature is 823 K, the composition z = 0.58

Drop solution enthalpies of Fe_3O_4 — Fe_2TiO_4 solid solutions in sodium molybdate at 973 K. The straight line connects the two end–members, the curve represents a second degree \times polynomial fit of the experimental data.



 Fe_3O_4 Fe_2TiO_4

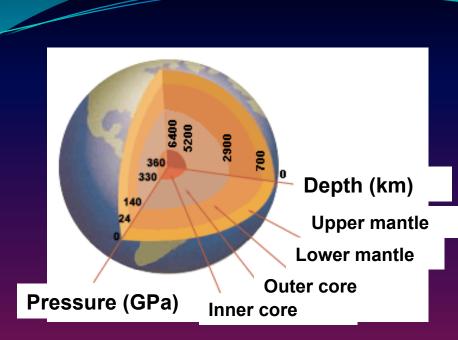
Events and Ambience 1980-95

The World

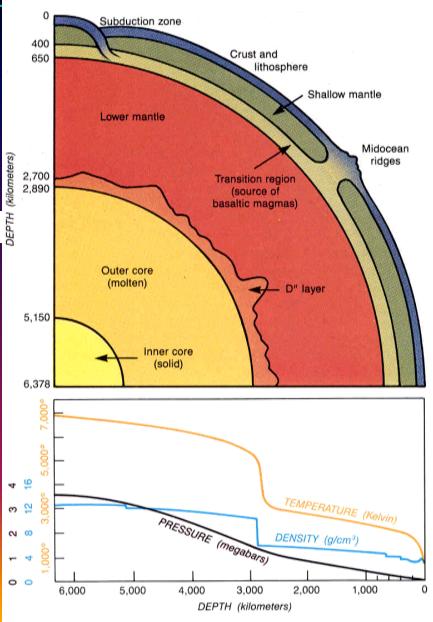
- Economic and political oscillations
- End of cold war
- Affirmative action programs
- Science funding variable

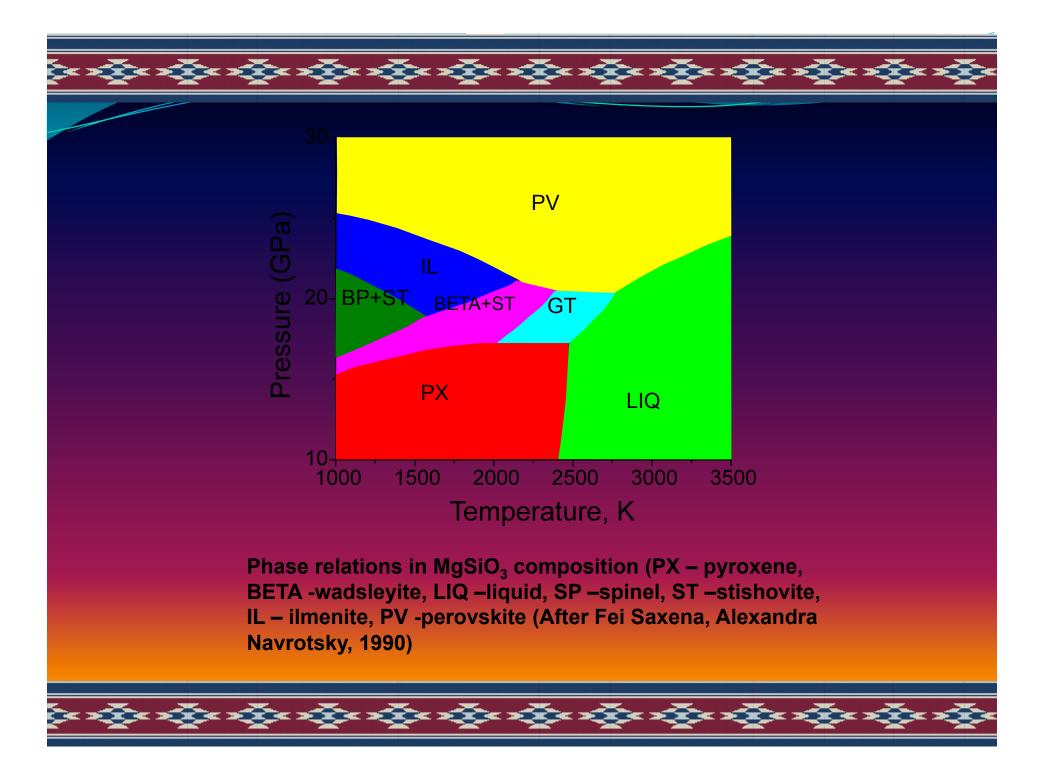
Alex

- Moved to Princeton 1985
- Research program
 - High pressure phases
 - Melts and glasses
 - Superconductors
 - Started work on zeolites
- Elected National Academy of Sciences 1993



Concentric shells of different phase assemblages with sharp discontinuities between them Olivine-spinelloid-spinel at 400 km Spinel- perovskite + periclase at 670 km Core-mantle boundary





Events and Ambience 1995-2005

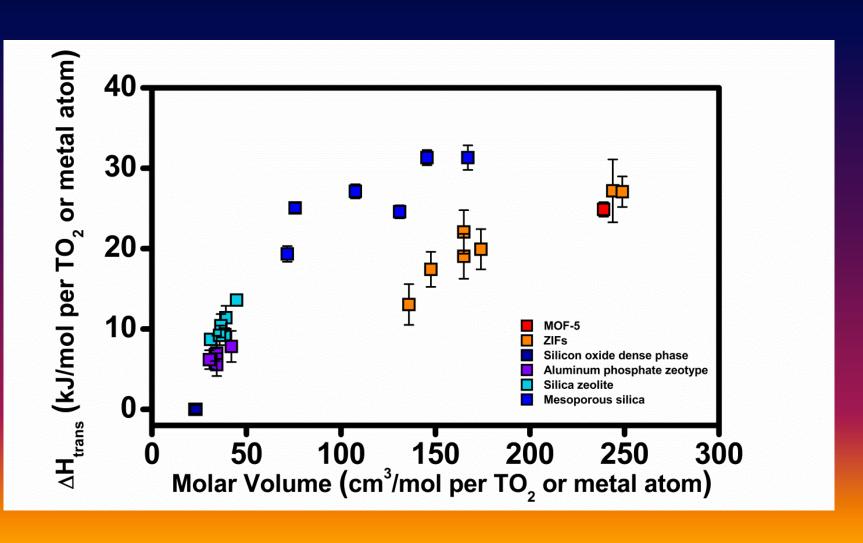
The World

- Economic and political oscillations
- 911 and increasing violence
- Science funding variable emphasis on nanotechnology
- Increasing diversity

Alex

- Moved to Davis 1997
- Benjamin Franklin Medal2002
- NEAT
- Research program
 - Zeolites
 - Nitrides
 - Ceramics and functional materials
 - Nanomaterials

Zeolite, Zeotype, and Metal-Organic Framework Energetics



Events and Ambience 2005-2016

The World

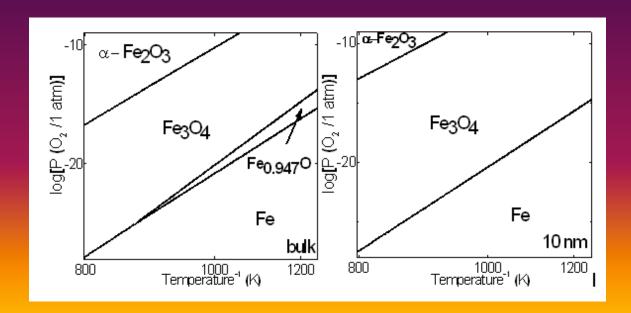
- Economic and political oscillations
- The recession and budget constraints at universities
- Science funding variable
 - Stimulus funding 2008-13
 - Major cutbacks 2012 ?
 - Emphasis on large projects
- Increasing diversity
- Increased university hiring

Alex

- Roebling Medal 2009
- Interim Dean of Mathematical and Physical Sciences 2013 present
- Goldschmidt Medal 2016
- Research program
 - Zeolites and MOFs
 - Nanomaterials
 - DOE EFRCs part of Materials Science of Actinides and 2 others
 - DOE Hub part of Critical Materials Institute

Oxidation-Reduction Equilibria among Transition Metal Oxides Change Dramatically at the Nanoscale Because of Differences in Spinels, M3O4 have lower surface energies than divalent oxides MO and Surface Energies 13O4, expanding the spinel stability field.

- For example, for no m iron oxides, wustite FeO has no stability field at all, with iron coexisting with magnetite
- Relevant to materials processing, environmental science, geology, and even biology
- Navrotsky et al. Science 330, 199-201 (2010)





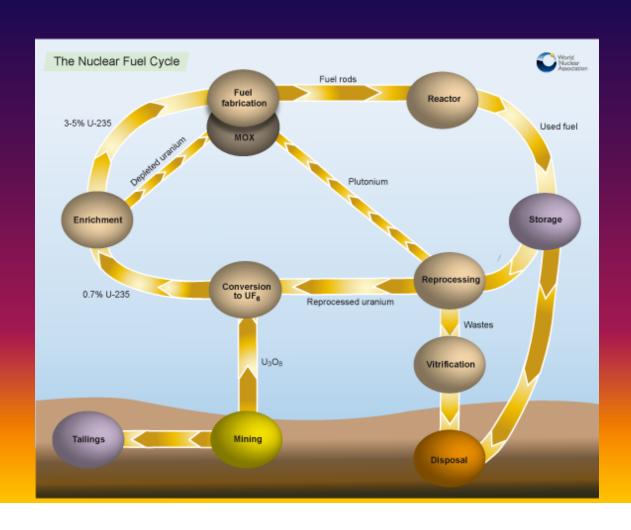
Major Current Projects

- Actinides
- Lanthanides and critical materials
- Fuel cell and battery materials
- Ultra high T processes
- Nanomaterials
- Open frameworks and confinement

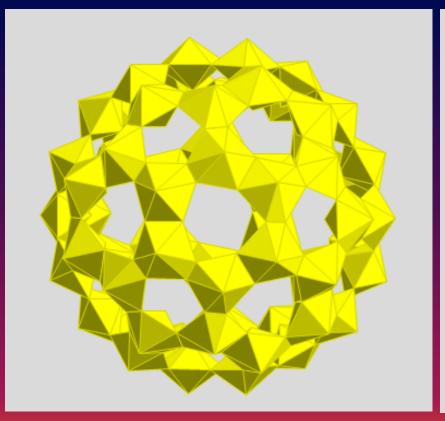


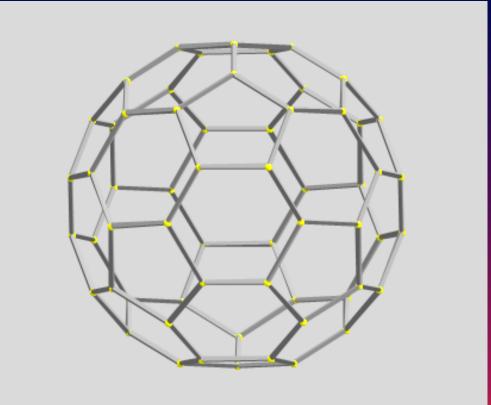
The Nuclear Fuel Cycle

Energy Frontier Research Center: Materials Science of Actinides



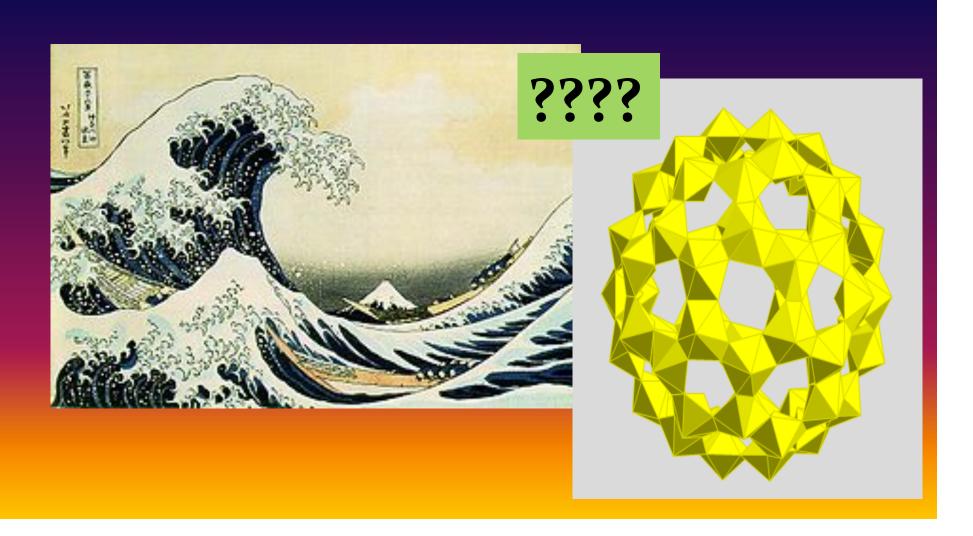
The Uranium Buckyball – U60





G. Sigmon et al., Angewandte Chemie 2009

Fukushima Nuclear Accident



Other Societally Important Issues

- Critical elements including rare earths
- Drought in the American West, including CA
- Food and sustainability and quality
- Catalysis
 - CO2 sequestration and reuse
 - Water splitting
- Sensors
 - Environmental
 - Medical diagnostics

Thermodynamics Rules

- Thermodynamics tells what is possible, kinetics how fast it will happen
- New processes and materials
 - More efficient technology
 - Environmental and social impacts
 - New materials and concerns have outrun the database
 - Know, optimize, do not compromise



Think Big